

5-Chloro-2-benzofuran-1,3-dione

Zai-Su Jin,^a Wen-Xi Wang,^a
Wei-Xiao Hu,^a Zhong-Yu Yang,^a
Hong-Wei Chen^b and
Ri-Cheng Xuan^{a*}

^aCollege of Pharmaceutical Sciences, Zhejiang University of Technology, Hangzhou 310023, People's Republic of China, and ^bCollege of Bioengineering, Changchun University of Technology, Changchun 130012, People's Republic of China

Correspondence e-mail: xuanrc@hotmail.com

Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$

R factor = 0.036

wR factor = 0.104

Data-to-parameter ratio = 15.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The asymmetric unit of the title compound, $\text{C}_8\text{H}_3\text{ClO}_3$, contains two molecules. The molecular packing is controlled by $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding.

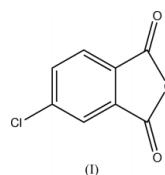
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Comment

The title compound, (I), is a valuable intermediate for Cu phthalocyanine and anthraquinone dyes (Bansho *et al.*, 1960), quinophthalone pigments (Dietmar *et al.*, 1998) and poly-ether-polyimide polymers (Brunelle *et al.*, 1999). Compound (I) can be made from phthalic anhydride by chlorination, but it is very difficult to separate (I) from the by-product 4,5-dichlorophthalic anhydride by recrystallization. This paper reports the structure analysis of (I).



The asymmetric unit contains two molecules, which are very similar (Fig. 1). These two molecules are essentially planar, making a dihedral angle of $58.13(5)^\circ$. Except for the $\text{C}-\text{Cl}$ bond lengths, which are 0.01 \AA longer than those observed in 4,5-dichlorophthalic anhydride (Ojala *et al.*, 1999), other bond lengths and angles are similar for the two compounds. The packing of the molecules in the crystal is governed by weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Fig. 2 and Table 2).

Experimental

The title compound was obtained by dehydration of 4-chlorophthalic acid, which was synthesized in our laboratory following the procedure described by Ayling (1929). Crystals suitable for X-ray analysis were obtained by slow evaporation of a saturated carbon tetrachloride solution at room temperature.

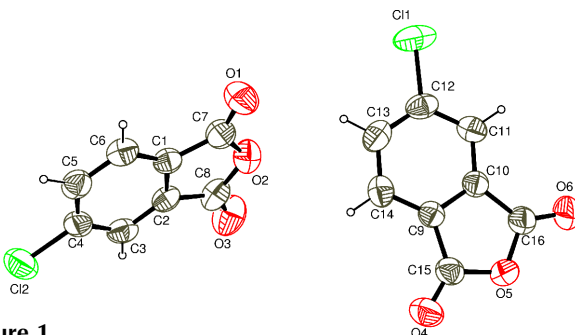


Figure 1

The asymmetric unit of (I), showing the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level for non-H atoms.

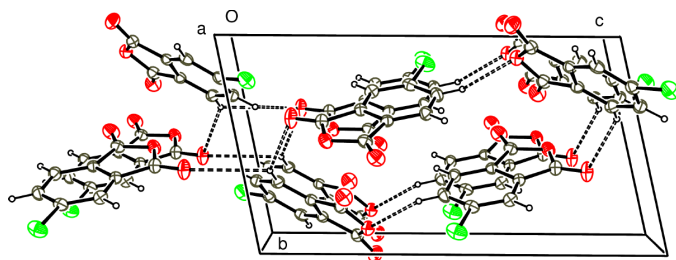


Figure 2
Packing diagram of (I), showing the hydrogen bonds as dashed lines.

Crystal data

$C_8H_5ClO_3$	$Z = 4$
$M_r = 182.55$	$D_x = 1.604 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 6.929 (2) \text{ \AA}$	Cell parameters from 4173 reflections
$b = 7.8540 (10) \text{ \AA}$	$\theta = 1.5\text{--}27.2^\circ$
$c = 14.262 (2) \text{ \AA}$	$\mu = 0.46 \text{ mm}^{-1}$
$\alpha = 77.69 (1)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 88.09 (1)^\circ$	Prism, colourless
$\gamma = 85.48 (2)^\circ$	$0.4 \times 0.2 \times 0.1 \text{ mm}$
$V = 755.8 (3) \text{ \AA}^3$	

Data collection

Enraf–Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.015$
$\omega/2\theta$ scans	$\theta_{\text{max}} = 27.2^\circ$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$h = -1 \rightarrow 8$
$T_{\text{min}} = 0.837$, $T_{\text{max}} = 0.955$	$k = -10 \rightarrow 10$
4173 measured reflections	$l = -18 \rightarrow 18$
3336 independent reflections	3 standard reflections every 100 reflections
2184 reflections with $I > 2\sigma(I)$	intensity decay: 0.5%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0367P)^2 + 0.2624P]$
$R[F^2 > 2\sigma(F^2)] = 0.036$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.104$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$
3336 reflections	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
217 parameters	
H-atom parameters constrained	

Table 1
Selected geometric parameters (\AA).

C1—C2	1.380 (3)	C9—C10	1.374 (3)
C1—C7	1.468 (3)	C9—C15	1.470 (3)
C2—C8	1.470 (3)	C10—C16	1.471 (3)
C4—C12	1.729 (2)	C12—C11	1.731 (2)
C7—O1	1.189 (3)	C15—O4	1.193 (3)
C7—O2	1.388 (3)	C15—O5	1.386 (3)
C8—O3	1.183 (3)	C16—O6	1.181 (3)
C8—O2	1.393 (3)	C16—O5	1.395 (3)

Table 2
Hydrogen-bonding geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C3—H3 \cdots O1 ⁱ	0.93	2.80	3.292 (3)	114
C6—H6 \cdots O6 ⁱⁱ	0.93	3.03	3.466 (3)	111
C6—H6 \cdots O6 ⁱⁱⁱ	0.93	2.89	3.690 (3)	145
C11—H11 \cdots O2 ^{iv}	0.93	2.94	3.675 (3)	137
C13—H13 \cdots O1	0.93	2.63	3.522 (3)	161
C13—H13 \cdots O2	0.93	2.81	3.453 (3)	127
C14—H14 \cdots O4 ^v	0.93	2.61	3.519 (3)	166

Symmetry codes: (i) $1 + x, y, z$; (ii) $x, y, 1 + z$; (iii) $-x, 1 - y, 1 - z$; (iv) $-x, -y, 1 - z$; (v) $1 - x, 1 - y, 1 - z$.

All H atoms were located in difference Fourier maps but were introduced in the refinement in calculated positions ($C\text{—}H = 0.93 \text{ \AA}$) and treated as riding, with $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994; cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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